

George Stokes  
(1819 - 1903)



Max Born  
(1882 - 1970)



E. Schrodinger  
(1887 - 1961)

### Historical Perspective

1853 Fluorescence described by Sir George G. Stokes

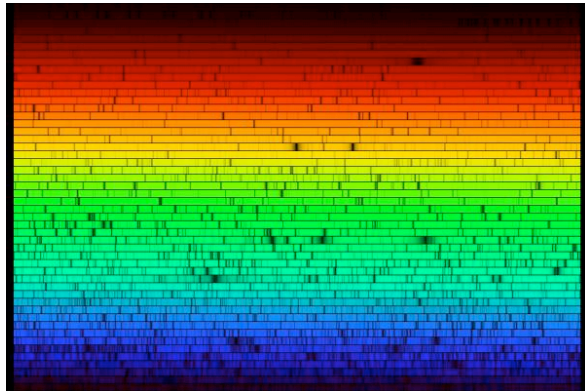
1913 Fluorescence microscope developed  
Heinrich Lehmann  
August Köhler & Carl Reichert (Zeiss)

1930s Secondary fluorescence technique  
Max Haitinger

1945 Quinine based fluorophore  
John Hershel

1950s Indirect Immunofluorescence  
Albert Coons & Nathan Kaplan

### Solar Spectrum



"Fraunhofer Lines"  
(1814)

Kurucz et al (1984)

### Visible Spectrum



### Solar Spectrum



H<sub>2</sub>

He<sub>2</sub>

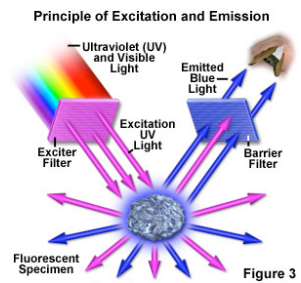
O<sub>2</sub>

C

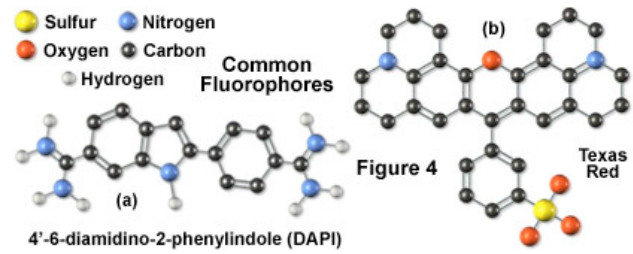
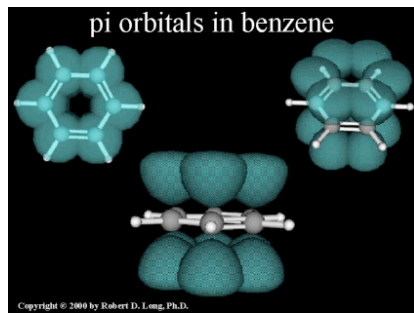
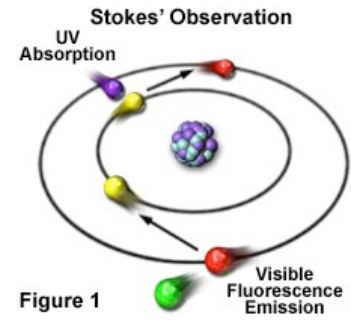
N<sub>2</sub>

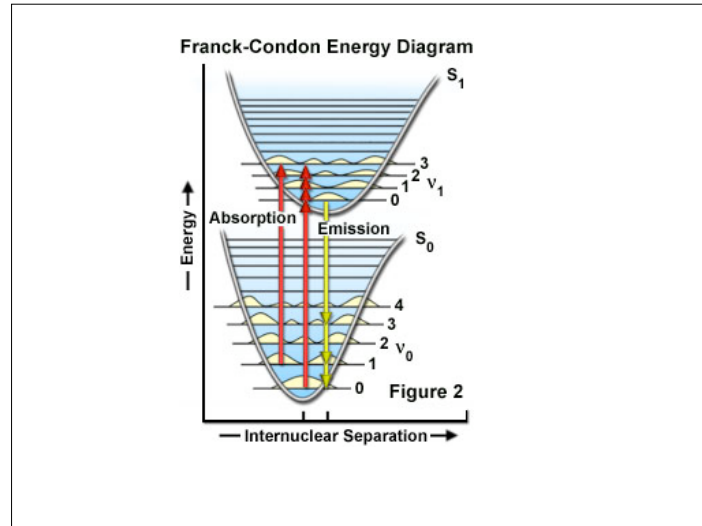
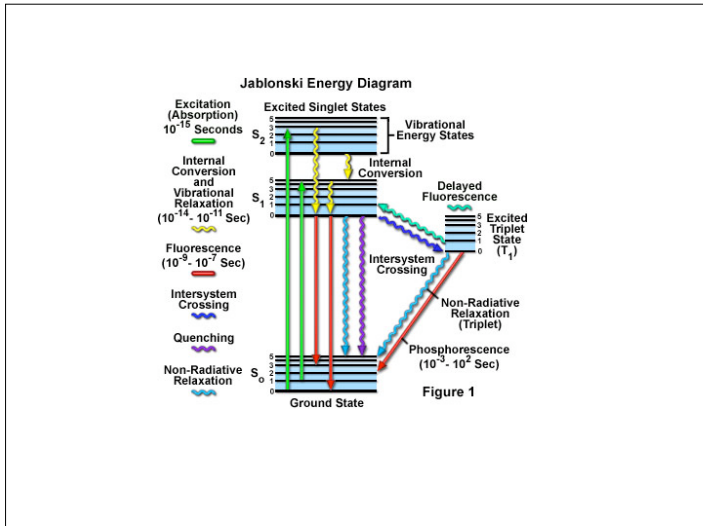
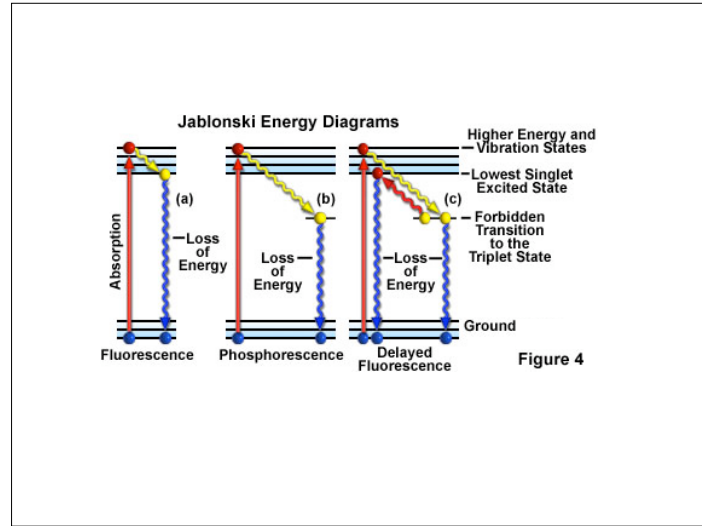
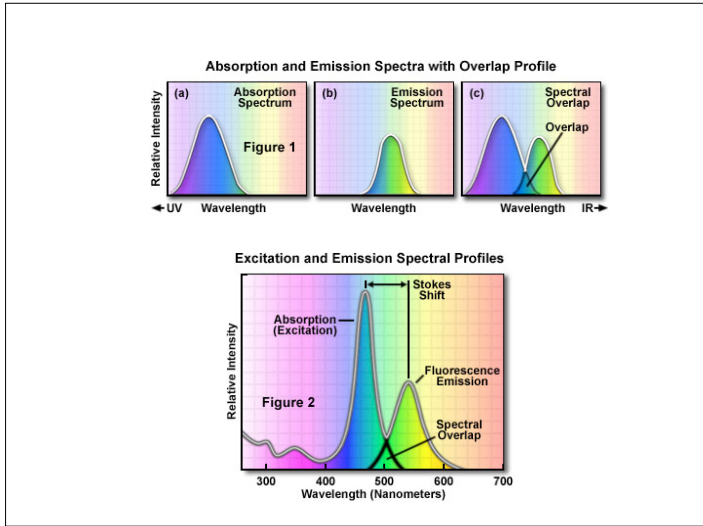
Si

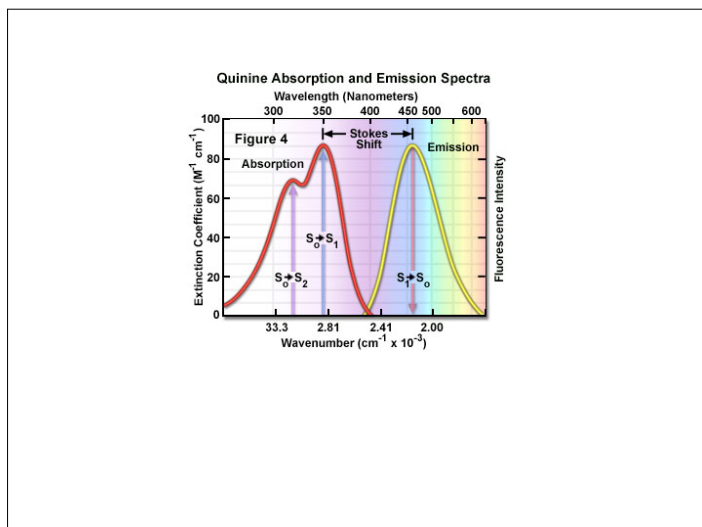
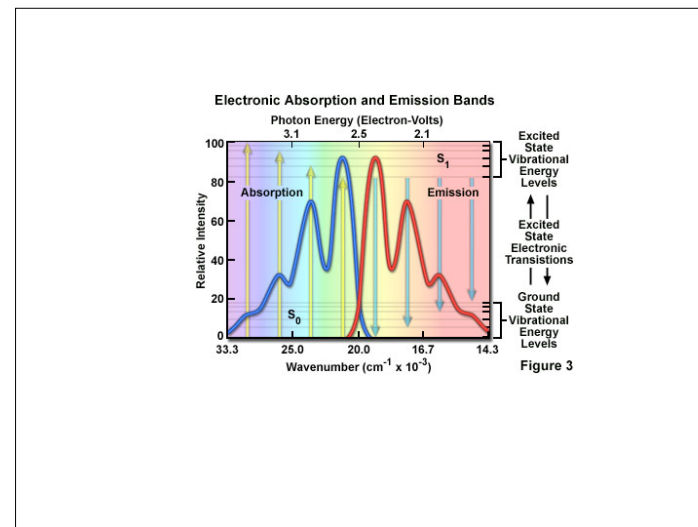
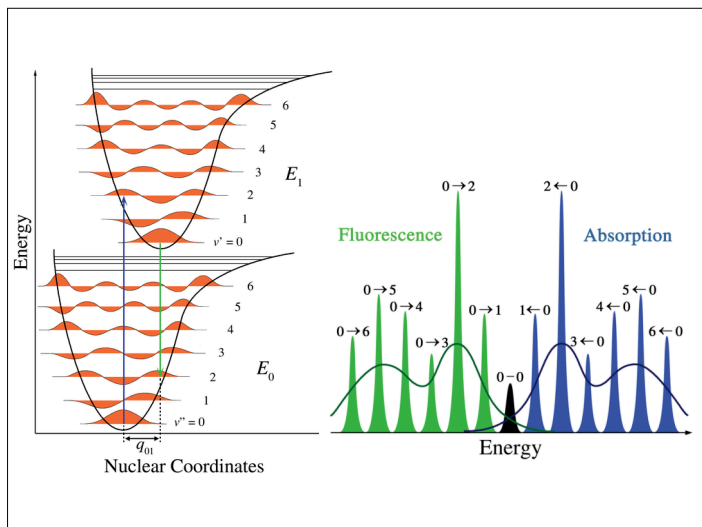
(Kirchhoff, Bunsen, others; 1860s)



Fluorspar







**Quantum Yield (photons emitted / photons absorbed)**

| <u>Dye</u>      | <u>Solvent</u> | <u>Exc</u> | <u>Em</u> | <u>QY</u> |
|-----------------|----------------|------------|-----------|-----------|
| Acridine Orange | Ethanol        | 493        | 535       | 0.46      |
| Benzene         | Ethanol        | 248        | 300-350   | 0.04      |
| Chlorophyll-A   | Ethanol        | 440        | 685       | 0.23      |
| Eosin           | Water          | 521        | 544       | 0.16      |
| Fluorescein     | Water          | 437        | 515       | 0.92      |
| Rhodamine-B     | Ethanol        | 555        | 627       | 0.97      |

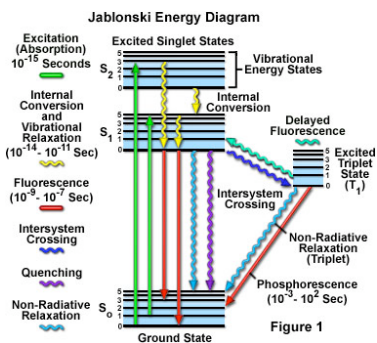
## Solvent Effects

Triplet Blinking

Fluorescence Quenching

Photobleaching

## Transition to the Dark Side

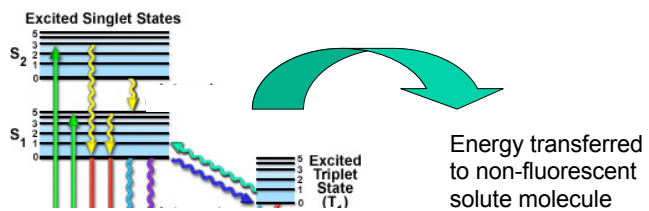


Triplet Blinking  
- stable triplet state  
- S<sub>1</sub> → T<sub>1</sub> due to solvent  
- halogens  
- transition metals

## Quenching

Collision with solutes

Resonance energy transfer



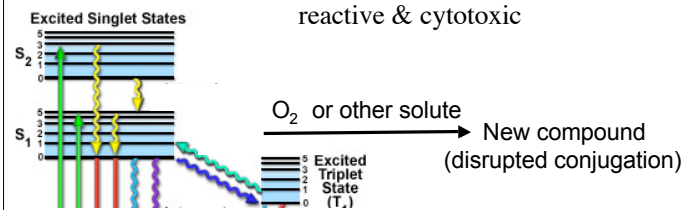
## Death of a Fluor

### Photobleaching

T<sub>1</sub> is highly reactive

O<sub>2</sub> likes T<sub>1</sub> electrons

Oxygen radicals are highly reactive & cytotoxic



Differential Photobleaching in Multiply-Stained Cell Cultures

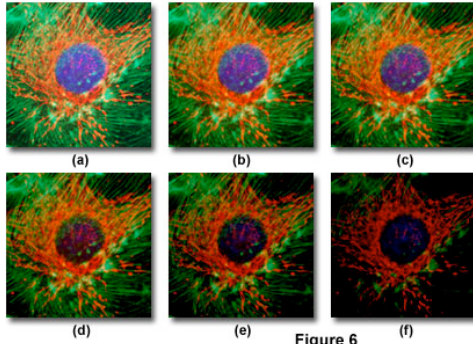


Figure 6

Differential Photobleaching in Multiply-Stained Tissues

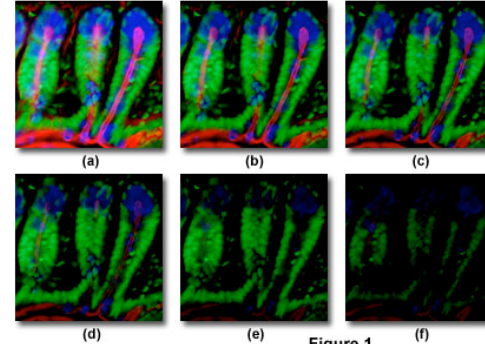


Figure 1

Reflected Light (Epifluorescence) Microscopy

Dichromatic Mirror Function in Reflected Light Fluorescence Illumination

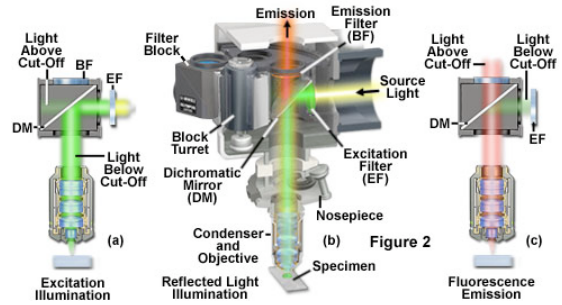


Figure 2

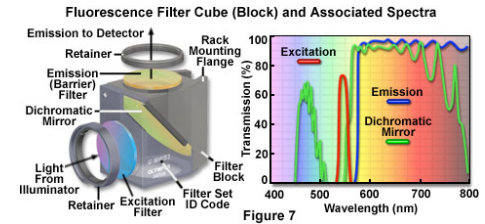
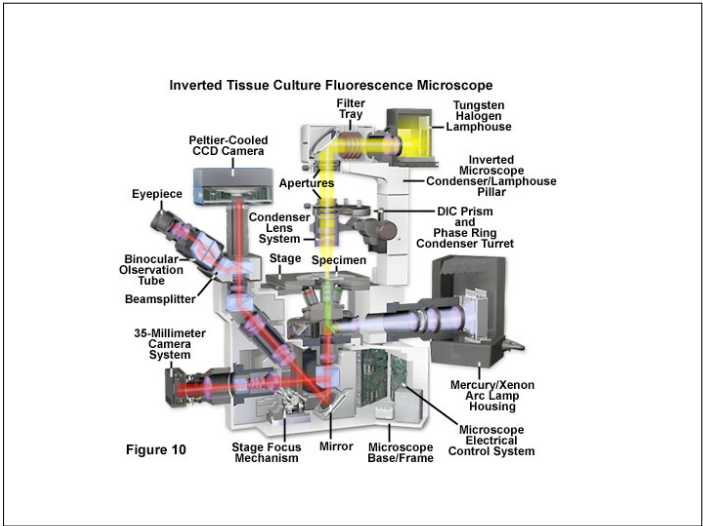
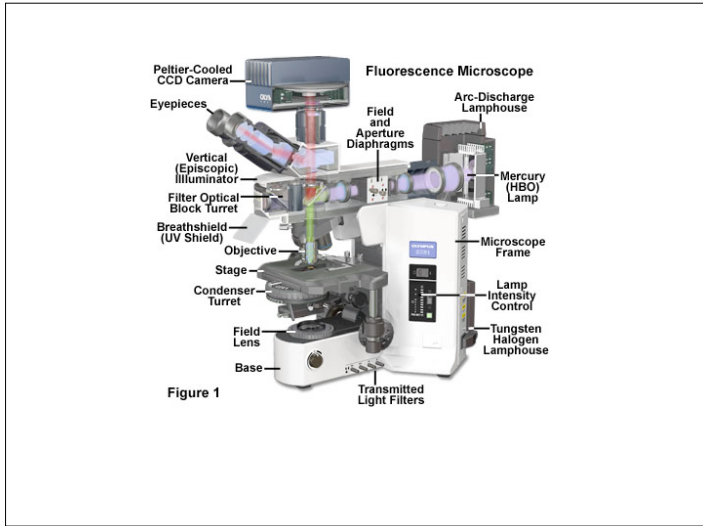
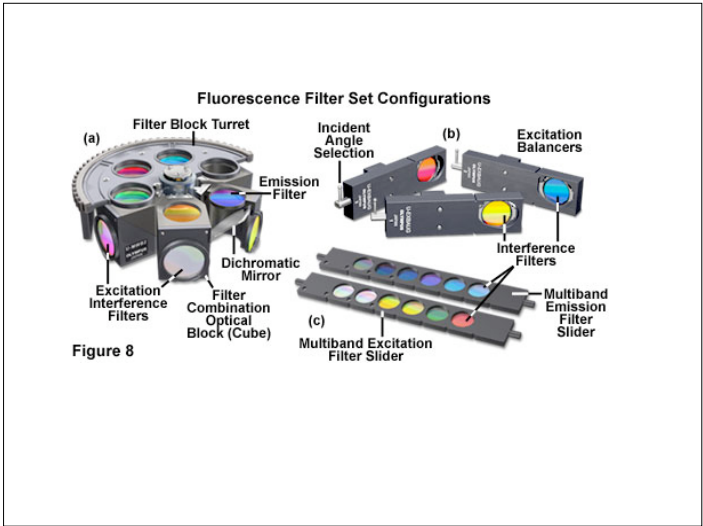
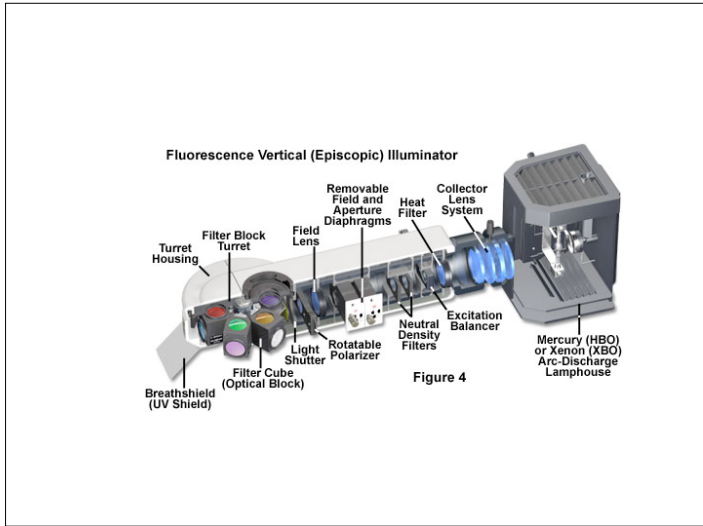
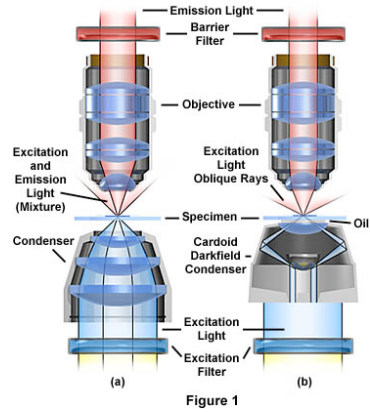


Figure 7

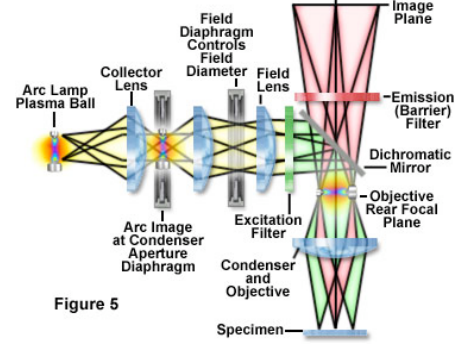




### Transmission Fluorescence Microscopy



### Köhler Illumination in Reflected Light Fluorescence



### Fluorescence Microscope Arc-Discharge Lamp Housing

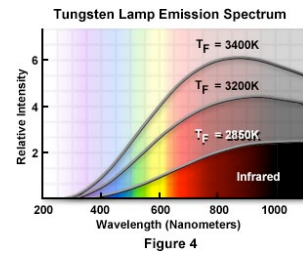
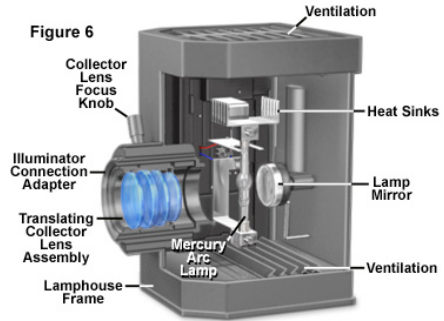






Figure 1

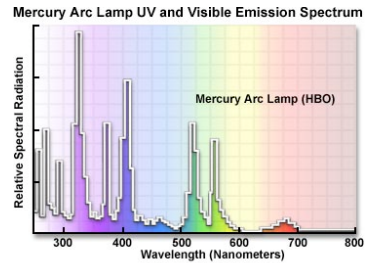


Figure 2

Xenon Arc Lamp Emission Spectrum

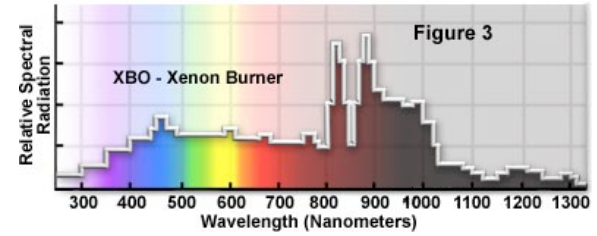


Figure 3

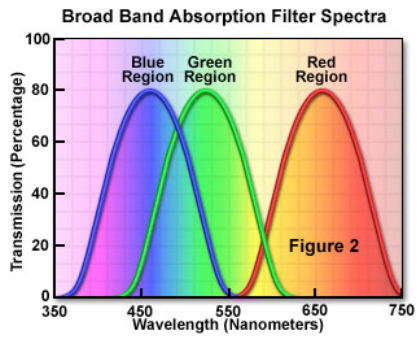


Figure 2

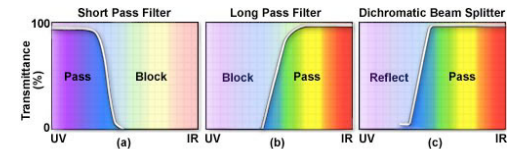


Figure 2

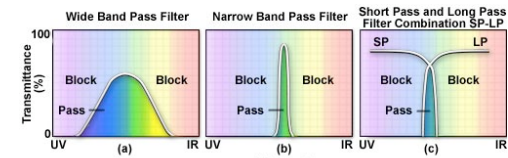
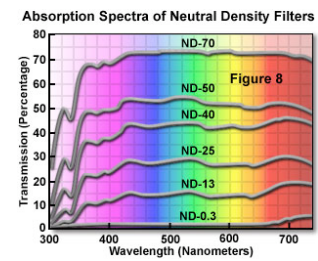
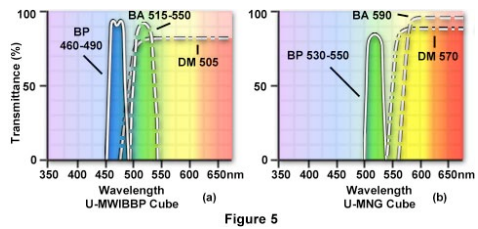
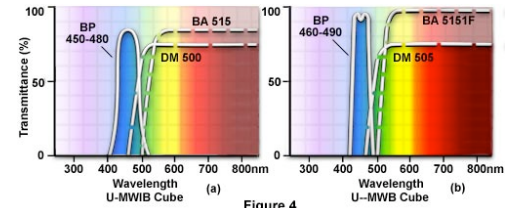
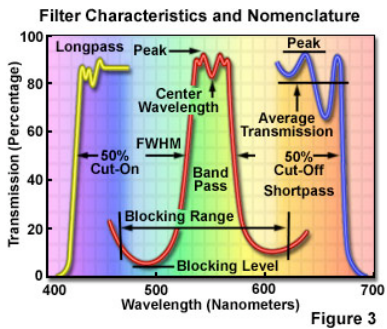
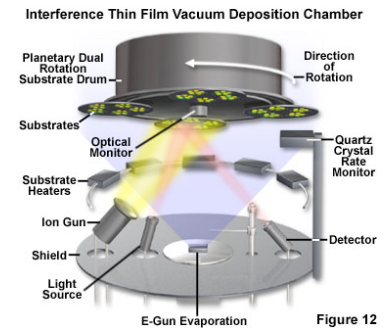
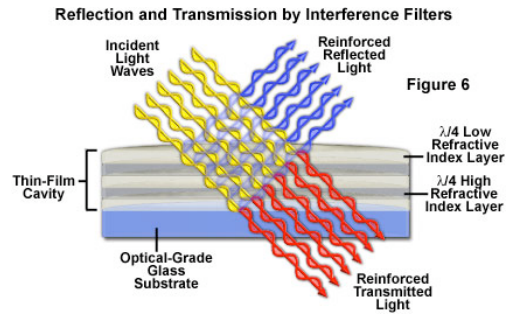
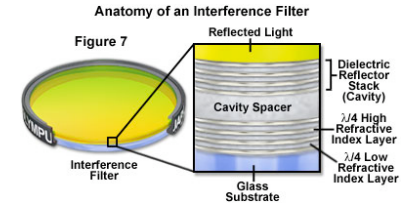
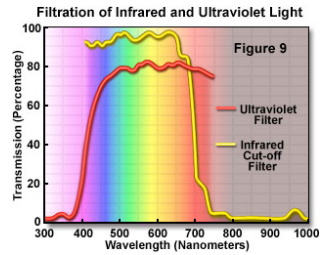
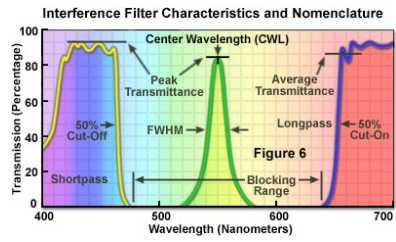


Figure 3

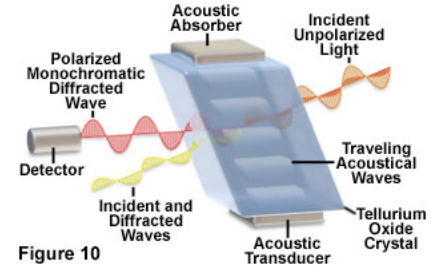




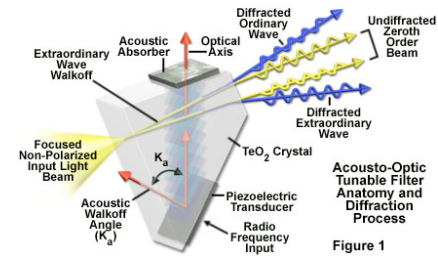
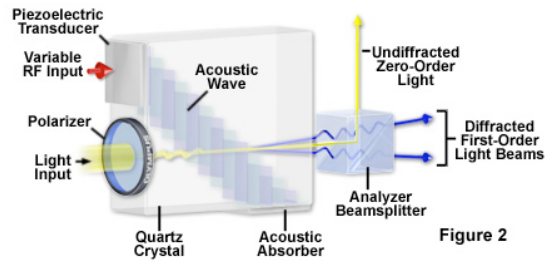


## AOTF

### Acousto-Optic Tunable Filter Anatomy



### Crystalline Quartz Collinear AOTF Configuration



## What to look for in a fluorophore

Fluorescence Spectrum

Quantum Yield

Extinction Coefficient

Stability (Photobleaching)

Sensitivity to Environment

Toxicity

Reactivity

Solubility