What to do with a fluorophore

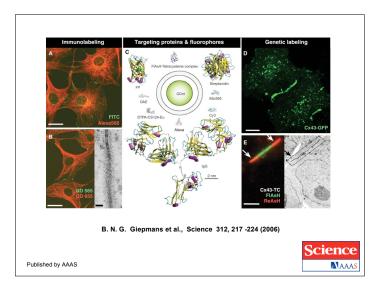
Intracellular localization (ER, Golgi, PM, nuclear, lysosome, MT, actin, ...)

<u>Dynamic processes</u> (protein synthesis, trafficking, turnover, DNA replication, cytoskeletal remodeling, membrane potential, enzyme activity, endocytosis, exocytosis, membrane fluidity)

<u>Cell signaling</u> (calcium, cyclic AMP/GMP, IP3, reactive oxygen species, pH,)

Cell integrity (live, dead, apoptotic)

What to look for in a fluorophore Fluorescence Spectrum Brightness ~ extinction coeff (10⁴ - 3x10⁵) x quantum yield (0.05 - 0.99) Stability (Photobleaching) Sensitivity to Environment (pH, metal ions, quenching agents) Toxicity Reactivity Solubility`



Types of fluorochromes

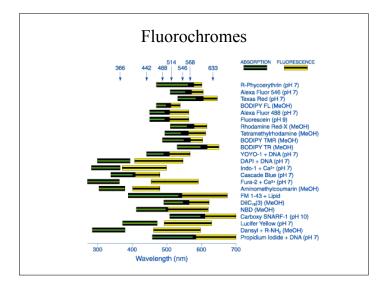
1. organic molecules (polyphenolic)

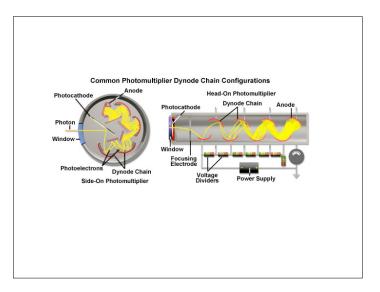
natural & synthetic

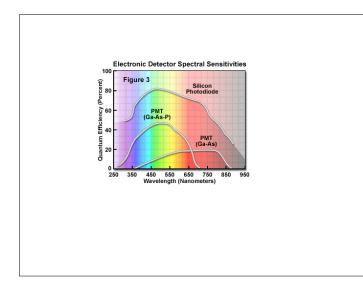
- 2. metal chelates (lanthanides)
- 3. semiconductor crystals (Q-dots)
- 4. fluorescent proteins

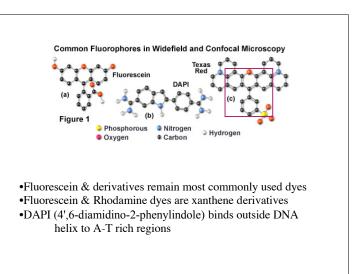
natural & engineered

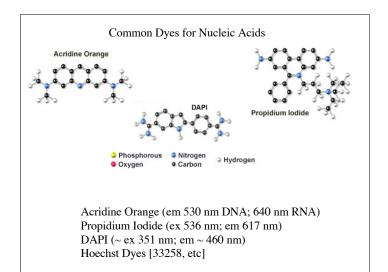
5. expressible affinity reagents

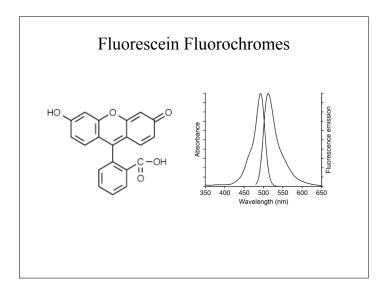


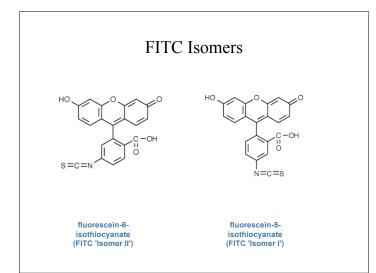


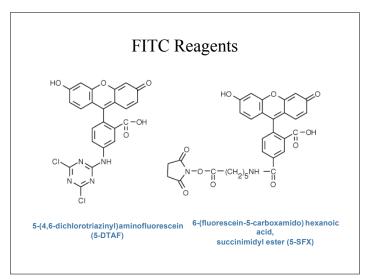


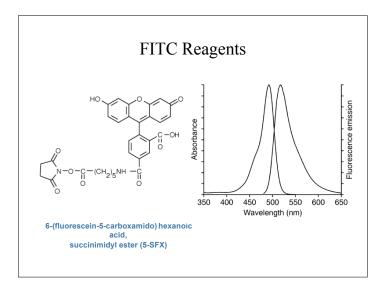


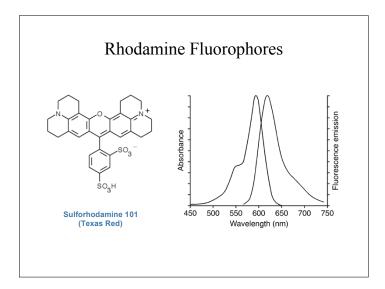


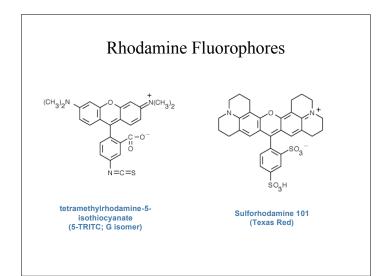


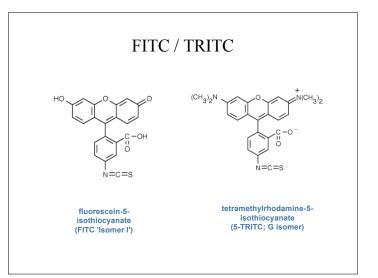


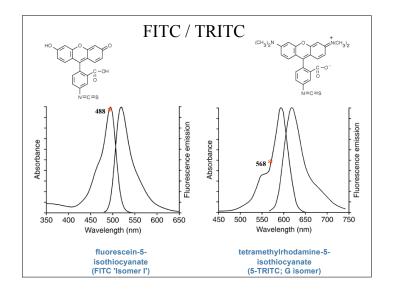


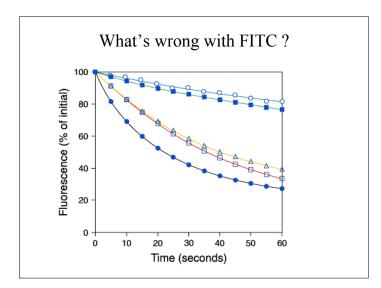


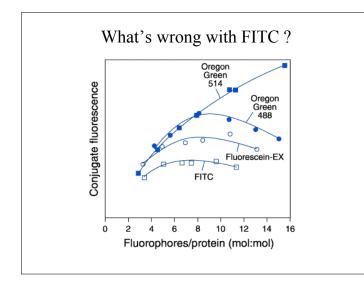


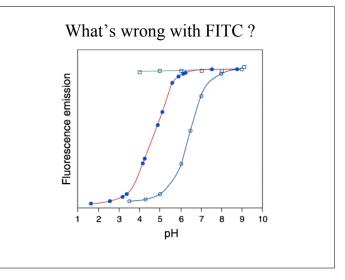


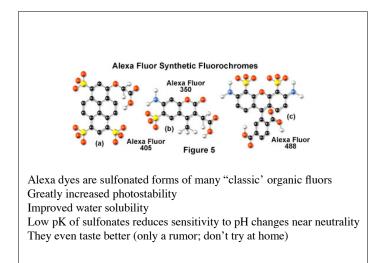


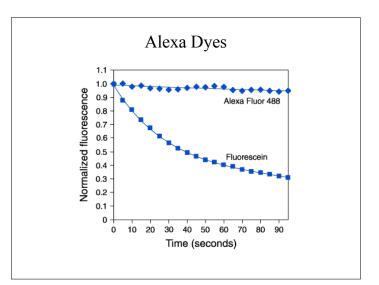


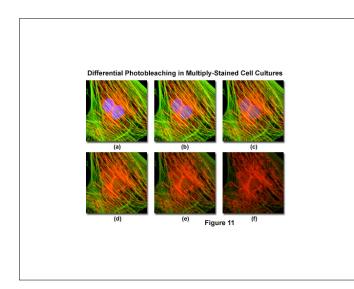


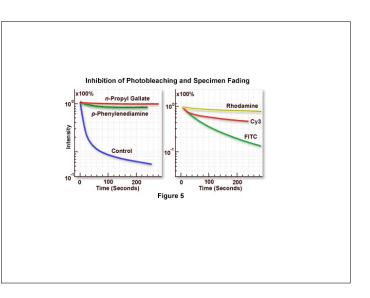


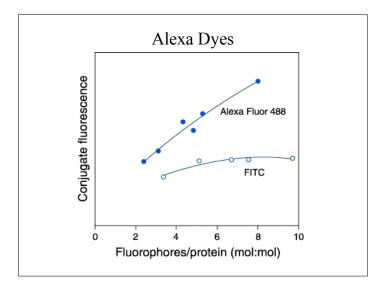


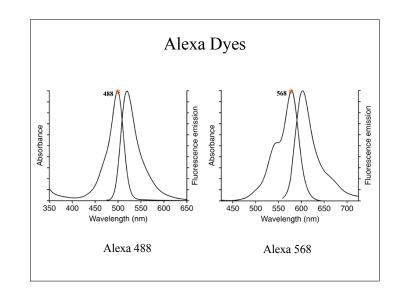


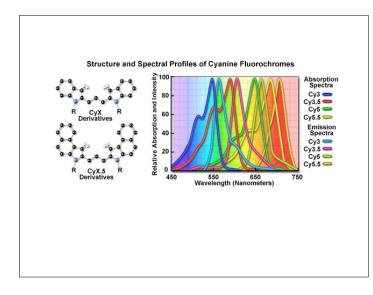


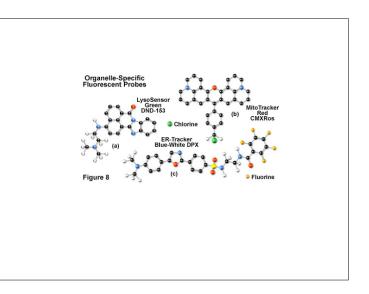


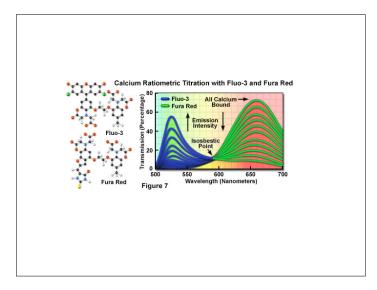




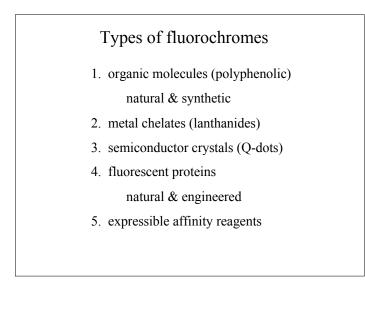






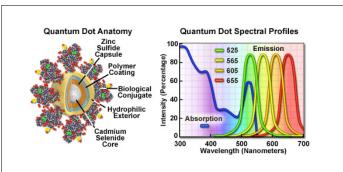


	1 1.01 3 Li 6.94 11 Na	2 4 Be 9.01 12 Mg		I	Pe Ele	em	of 1 en		13 5 B 10.81 13 Al	14 6 C 12.01 14 Si	15 7 N 14.01 15 P	16 8 15.99 16 S	17 9 F 19.00 17 C	18 2 He 4.00 10 Ne 20.18 18 Ar				
	19 K	24.31 20 Ca	3 21 Sc	4 22 Ti	5 23 V	24 Cr	7 25 Mn	26 Fe	9 27 Co	10 28 Ni	29 Cu	12 30 Zn	26.98 31	^{28.09}	30.97 33 As	32.07 34 Se	35.45 35	^{39.95} 36 Kr
	39.10	40.08	44.96	47.87	50.94	52.00	54.94	55.85	58.93	58.69	Cu 63.55	65.41	Ga 69.72	Ge 72.64	74.92	78.96	79.90	83.80
	37 Rb	38 Sr	39 Y	⁴⁰ Zr	Nb	42 Mo	43 TC	Ru Ru	⁴⁵ Rh	Pd	Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
	85.47	87.62	88.91	91.22	92.91	95.94	(98)	101.07	102.91	105.42	107.87	112.41	114.82	118.71	121.76	127.60	126.90	131.29
	55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	⁷⁵ Re	76 Os	Ir	78 Pt	⁷⁹ Au	80 H C	TI TI	Pb	⁸³ Bi	Po	At	86 Rn
	132.91 87	137.33 88	138.91 89	178.49	180.95	183.84	186.21	190.23	192.22	195.08	196.97	200.59	204.38	207.2	208.98	(209)	(210)	(222)
	67 Fr (223)	88 Ra (226)	AC (227)	104 Rf (261)	Db (262)	Sg (266)	Bh (264)	Hs (270)	109 Mt (268)	Ds (281)	Rg (272)							
				_	Ce	59 Pr	60 Nd	61 Pm	Sm	Eu Eu	64 Gd	Tb	66 Dy	67 Ho	68 Er	Tm	70 Yb	71 Lu
	Molecular Research Institute					140.91	144.24	(145)	150.36	151.97	157.25	158.93	162.50	164.93	167.26	168.93	173.04	174.97
						91 Pa 231.04	92 U 238.03	93 Np	94 Pu (244)	95 Am	96 Cm (247)	97 Bk (247)	98 Cf (251)	99 Es (252)	100 Fm (257)	101 Md (258)	102 No (259)	103 Lr (262)

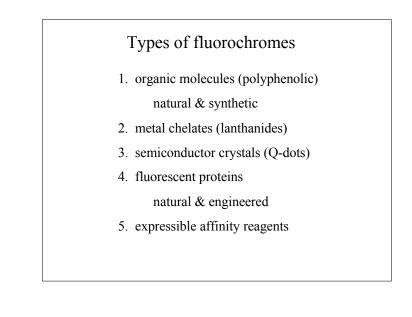


Types of fluorochromes

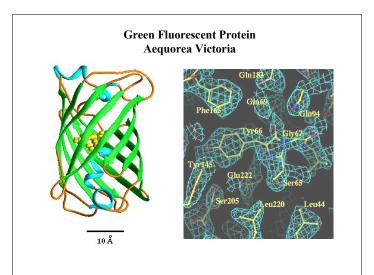
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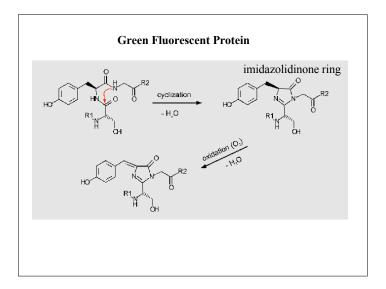


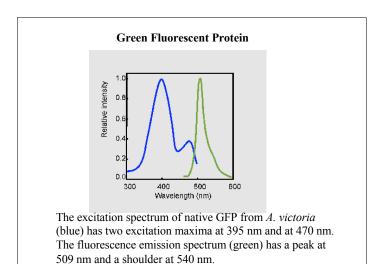
Fluorescence emission peak increases with size Can activate all colors with same excitation band Brightness increases with shorter wavelength excitation ZnS coating increases photostability (huge !!) Polymer coating is necessary to improve hydrophilicity Disadvantage = size (10 - 15 nm)

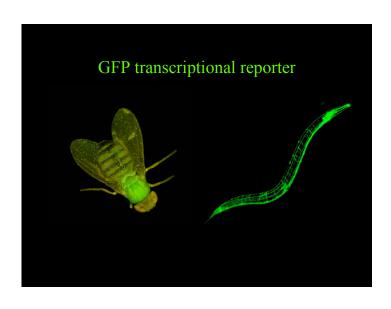




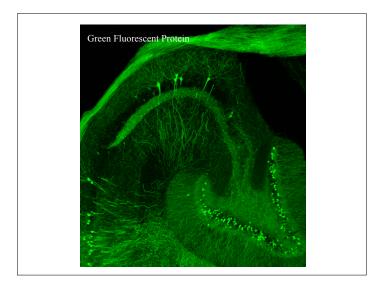


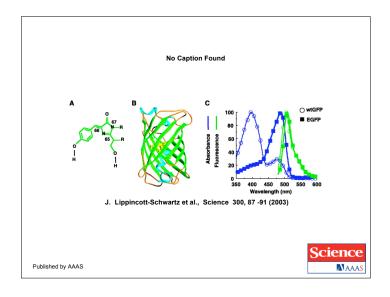


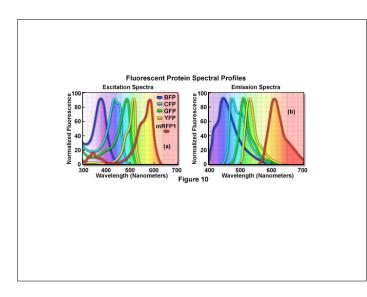


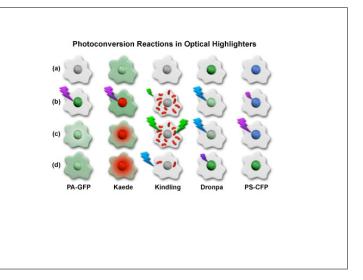


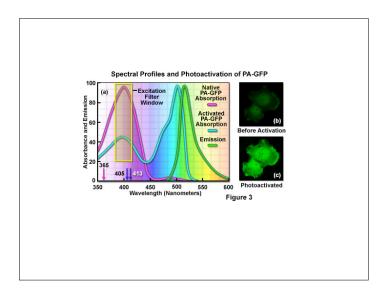


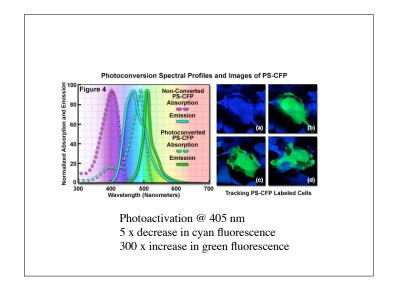


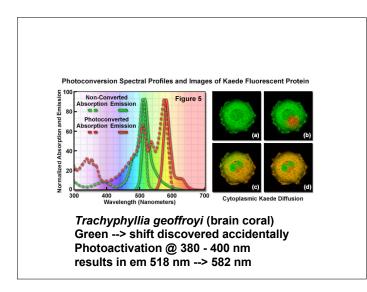


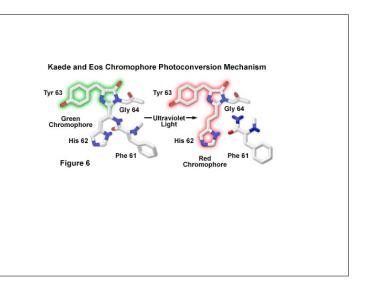


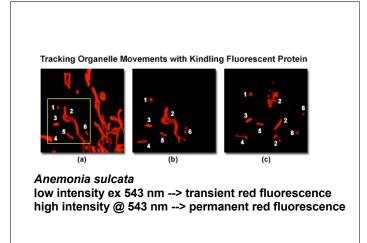


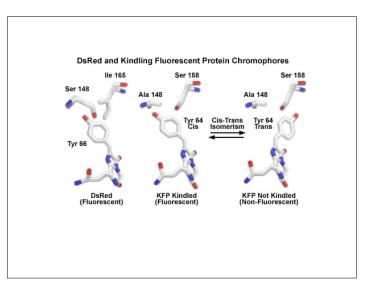


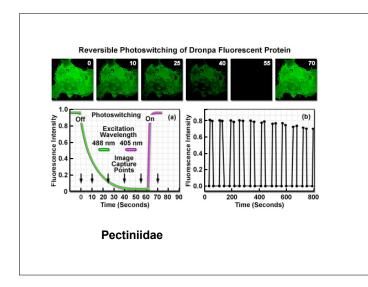


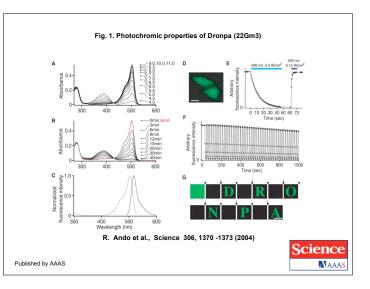


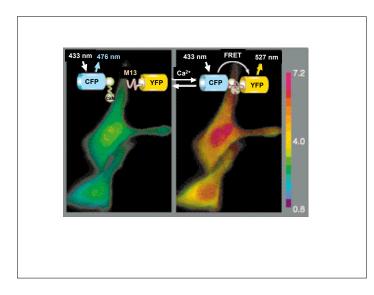


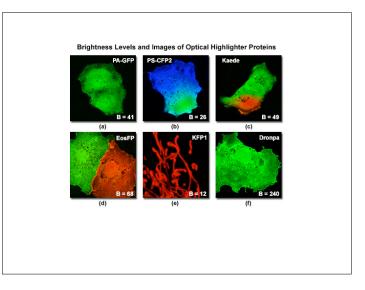












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